

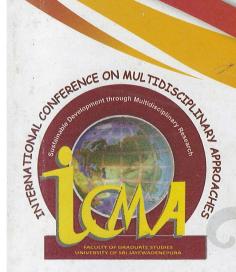




4th International Conference on Multidisciplinary Approaches - 2017

"Sustainable Development through Multidisciplinary Research"

PROCEEDINGS



JOINTLY ORGANIZED BY

FACULTY OF GRADUATE STUDIES
UNIVERSITY OF SRI JAYEWARDENEPURA

MINISTRY OF SCIENCE, TECHNOLOGY AND RESEARCH

NATIONAL SCIENCE FOUNDATION

20TH - 22ND OF SEPTEMBER, 2017

Nugegoda, Sri Lanka

Proceedings of the 4^{th} International Conference on Multidisciplinary Approaches -2017

Faculty of Graduate Studies University f Sri Jayewardenepura Nugegoda Sri Lanka

208 Pages

ISSN: 2386 - 1509

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Published by: Faculty of Graduate Studies, University of Sri Jayewardenepura, Nugegoda, Sri Lanka

Tel: +94 112881571 Fax: +94 112802551 Email: icma@sjp.ac.lk

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GREENER APPROACH TO REMOVE METHYLENE BLUE (MB) USING CALCIUM HYDROXIDE DERIVED FROM EGGSHELLS

Amarasignhe L.M.A.S. ¹, Wanniarachchi W.K.I.L. ² and Wanniarachchi D.D.C. ^{3*}

¹ Department of Science and Technology, Uva Wellassa University, Badulla

² Department of Physics, Center for Advanced Materials Research (CAMR), University of Sri Jayewardenepura, Sri Lanka

³ Central Instrument Unit, Faculty of Applied Sciences, University of Sri Jayewardenepura, Sri Lanka

The use of raw egg shells and egg shell powder calcined at 900°C as effective adsorbent materials for removal of dyes was studied. In order to understand the adsorption nature of these materials, methylene blue (MB) was used as the model dye. Though MB adsorption on to raw egg shell powder was previously recorded, a comparison on effective removal of dye using egg shell powder calcined at high temperature was not investigated. The two materials are inherently different as raw egg shell powder is mainly calcium carbonate while calcined egg shell powder is mostly calcium hydroxide. The kinetics of dye adsorption, type of adsorption mechanism and effect of light on adsorption process were compared for raw egg shell powder and calcined egg shell powder with sample containing dye only as the control. The results clearly showed that the enhanced efficiency of use of egg shell powder calcined at 900°C over raw egg shell powder for removing MB from water. The reported efficiency for calcined egg shell powder was 83% after two hours while raw egg shell powder indicated efficiency was 21% during the same period. Kinetics of MB dye removal follows a pseudo second order kinetics with rate constant of 30 min⁻¹. Adsorption properties of MB on to calcined egg shells indicate type II adsorption isotherm. In conclusion, the study showed use of calcined egg powder is a greener approach as waste is reused to remove dye.

Keywords: Egg-shell powder, Dye Removal, Methylene Blue, Eco friendly

Acknowledgements: Authors wish to thank Department of Science and Technology, Department of Animal Science, UWU for use of lab facilities for this study. AMRC at University of Sri Jayewardenepura is acknowledged for the for the XRD study.

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